

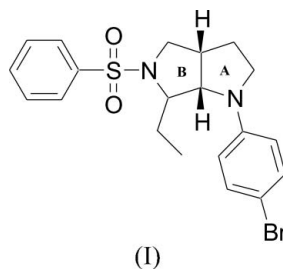
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Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(C-C) = 0.004$ Å
R factor = 0.035
 wR factor = 0.088
Data-to-parameter ratio = 18.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**cis-1-(4-Bromophenyl)-6-ethyl-5-(phenylsulfonyl)-perhydropyrrolo[3,4-*b*]pyrrole**In the title compound, $C_{20}H_{23}BrN_2O_2S$, both pyrrolidine rings adopt twist conformations. The crystal packing is stabilized by intermolecular $C-H \cdots O$ hydrogen bonds and $\pi-\pi$ interactions.Received 21 March 2007
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Comment

Pyrrolopyrrole compounds exhibit anti-inflammatory and analgesic activities (Rooks *et al.*, 1982; Muchowski *et al.*, 1989). Inhibitors of human cytomegalovirus (HCMV) protease have been designed based on the 5-oxo-hexahydropyrrolo[3,2-*b*]pyrrole ring system (Borthwick *et al.*, 2000). It has been shown that *N*-substituted pyrrole derivatives inhibit human immunodeficiency virus type-1 (HIV-1) (Jiang *et al.*, 2004). These derivatives also possess antileukemic activity (Anderson & Mach, 1987) and some of them are used as platelet activating factor (PAF) antagonists (Weissman *et al.*, 1993; Le Naour *et al.*, 1994). In view of this biological importance, the crystal structure of the title compound, (I), has been determined and the results are presented here.Bond lengths and angles in (I) (Fig. 1) agree with those observed in a similar structure, *cis*-1-(4-bromophenyl)-6-ethyl-5-tosylperhydropyrrolo[3,4-*b*]pyrrole (Chinnakali *et al.*, 2007). The sums of the bond angles around atoms N1 (357.3°) and N2 (352.1°) indicate sp^2 -hybridization. Atom S1 has a distorted tetrahedral geometry, with angles O1–S1–O2 [120.06 (14°)] and N2–S1–O2 [106.52 (13°)] deviating significantly from the ideal value, as a result of the Thorpe–Ingold effect (Bassindale, 1984).Both pyrrolidine rings A (N1/C2–C4/C7) and B (N2/C4–C7) adopt twist conformations. The puckering parameters (Cremer & Pople, 1975) and the asymmetry parameters (Nardelli, 1983) are $q_2 = 0.320$ (3) Å and $\varphi = 87.2$ (5°) and $\Delta C_2(N1) = 3.1$ (3°) for ring A, and $q_2 = 0.281$ (3) Å, $\varphi = 302.4$ (5°) and $\Delta C_2(C5) = 1.7$ (3°) for ring B. The two aromatic rings are almost parallel to one another with a dihedral angle of 4.9 (2°).In the crystal structure, $C-H \cdots O$ intermolecular hydrogen bonds link the molecules into a three-dimensional framework

(Table 1). In addition, a π - π interaction involving the C8–C13 ring of the molecule at (x, y, z) and the C14–C19 ring of the molecule at $(-1+x, y, z)$ is observed, with a centroid–centroid distance of 3.632 (2) Å.

Experimental

A mixture of 2-(*N*-allyl-*N*-tosylamino)butanal (1 mmol) and 2-(*p*-bromo)phenylthiazolidine-4-carboxylic acid (1 mmol) in toluene (20 ml) was refluxed until the disappearance of the starting materials, as evidenced by thin-layer chromatography. The solvent was evaporated under vacuum and the residue was then column-chromatographed with a hexane–ethyl acetate mixture (8:2) to obtain the title compound.

Crystal data

$C_{20}H_{23}BrN_2O_2S$	$V = 1948.03 (9) \text{ \AA}^3$
$M_r = 435.37$	$Z = 4$
Orthorhombic, $Pna2_1$	Mo $K\alpha$ radiation
$a = 7.9894 (2) \text{ \AA}$	$\mu = 2.23 \text{ mm}^{-1}$
$b = 18.8355 (5) \text{ \AA}$	$T = 293 (2) \text{ K}$
$c = 12.9451 (4) \text{ \AA}$	$0.25 \times 0.17 \times 0.15 \text{ mm}$

Data collection

Bruker Kappa APEX2 area-detector diffractometer	16756 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	4421 independent reflections
$T_{\min} = 0.640$, $T_{\max} = 0.715$	3229 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.028$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$	H-atom parameters constrained
$wR(F^2) = 0.088$	$\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$
$S = 1.02$	$\Delta\rho_{\text{min}} = -0.36 \text{ e \AA}^{-3}$
4421 reflections	Absolute structure: Flack (1983),
235 parameters	2097 Friedel pairs
1 restraint	Flack parameter: 0.003 (7)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C16-H16\cdots O1^i$	0.93	2.53	3.433 (4)	164
$C20-H20A\cdots O2^{ii}$	0.97	2.57	3.453 (4)	151

Symmetry codes: (i) $-x+3, -y, z+\frac{1}{2}$; (ii) $x-\frac{1}{2}, -y-\frac{1}{2}, z$.

H atoms were placed in idealized positions and allowed to ride on their parent atoms, with $C-H = 0.93\text{--}0.98 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$.

Data collection: APEX2 (Bruker, 2004); cell refinement: APEX2; data reduction: SAINT (Bruker, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: PLATON (Spek, 2003).

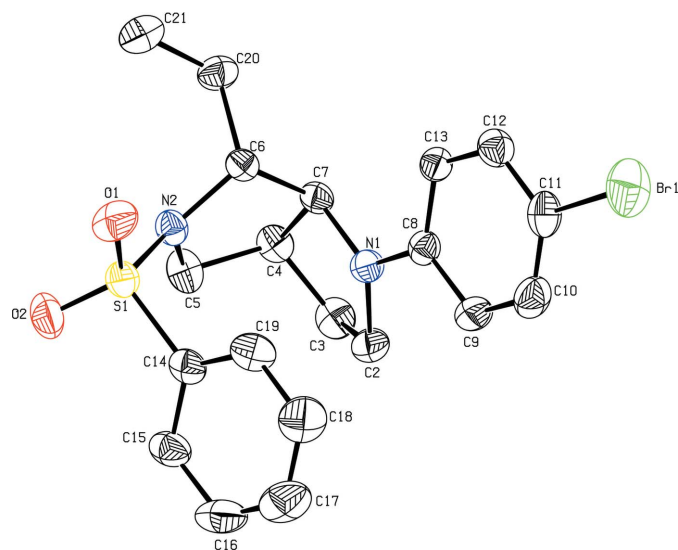


Figure 1

The molecular structure of (I), showing 30% probability displacement ellipsoids. H atoms have been omitted.

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References

- Anderson, W. K. & Mach, R. H. (1987). *J. Med. Chem.* **30**, 2109–2115.
- Bassindale, A. (1984). *The Third Dimension of Organic Chemistry*, ch. 1, p. 11. New York: John Wiley and Sons.
- Borthwick, A. D., Angier, S. J., Crame, A. J., Exall, A. M., Haley, T. M., Hart, G. J., Mason, A. M., Pennell, A. M. K. & Weingarten, G. G. (2000). *J. Med. Chem.* **43**, 4452–4464.
- Bruker (2004). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chinnakali, K., Poornachandran, M., Raghunathan, R. & Fun, H.-K. (2007). *Acta Cryst.* **E63**, o650–o651.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Jiang, S., Lu, H., Liu, S., Zhao, Q., He, Y. & Debnath, A. K. (2004). *Antimicrob. Agents Chemother.* **48**, 4349–4359.
- Le Naour, R., Clayette, P., Henin, Y., Mabondzo, A., Raoul, H., Bousseau, A. & Dormont, D. (1994). *J. Gen. Virol.* **75**, 1379–1388.
- Muchowski, J. M., Galeazzi, E., Greenhouse, R., Guzman, A., Perez, V., Ackerman, N., Ballaron, S. A., Rovito, J. R., Tomolonis, A. J. & Young, J. M. (1989). *J. Med. Chem.* **32**, 1202–1207.
- Nardelli, M. (1983). *Acta Cryst.* **C39**, 1141–1142.
- Rooks, W. H., Tomolonis, A. J., Maloney, P. J., Walloch, M. B. & Schuler, M. E. (1982). *Agents Actions*, **12**, 684–690.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Weissman, D., Poli, G., Bousseau, A. & Fauci, A. S. (1993). *Proc. Natl Acad. Sci. USA*, **90**, 2537–2541.